

APPENDIX E

EMANATION OF GASEOUS RADIONUCLIDES FROM THE SOIL SURFACE

This page intentionally left blank.

EMANATION OF GASEOUS RADIONUCLIDES FROM THE SOIL SURFACE

This section describes the migration of gaseous radionuclides from the buried waste, through the facility cover to the ground surface, and into the atmosphere. Then the consequences of the air emission are estimated. The gaseous radionuclides are tritium (as water vapor), ^{14}C (as carbon dioxide), and ^{222}Rn (an inert gas).

E.1. DIFFUSION FLUX AT THE GROUND SURFACE

For the discussion in this section, the waste is separated into solids and a gas-filled pore space. The solids contain the radioactive inventory, while the pore space collects any gaseous nuclides that may be released from the solids. Vitrified waste is expected to release gases very slowly, while low-level waste could release the gaseous radionuclides rapidly. The quantity of gaseous radionuclides in the waste solids decreases with time by radioactive decay, and by release from the waste solids into the waste pore space. It will be assumed that over the course of one year both the inventory in the waste solids and the fractional release rate are relatively constant. The rate at which the gaseous nuclides are released from the waste solids is calculated using the formula below.

$$\text{Gas Release Rate} = \lambda_C Q_S$$

where

- Q_S = quantity of a gaseous radionuclide in the waste solids at some time after closure, in Ci
- λ_C = fraction of the total released from the waste solids into the waste pore space per year. This fraction may vary slowly from year to year.

The gaseous radionuclides released from the solids then diffuse from the waste into the soil surrounding the disposal site. If this diffusion is slow, then the gaseous nuclides accumulate in the pore space until the diffusion rate away from the waste matches the release rate from the solids. If this diffusion is rapid, then the concentration in the waste pore space will be very small, but will be proportional to the concentration in the waste solids.

The diffusion of radioactive gases can be represented using Fick's law of diffusion with a loss term for radioactive decay (Jury 1991 Chapter 6). The ^{222}Rn will be discussed separately from the ^3H and ^{14}C because it is produced by a decay chain that includes ^{238}U , ^{234}U , ^{230}Th , and ^{226}Ra . For calculating diffusion in the presence of radioactive decay the source concentration is assumed to be constant because it changes very slowly with time compared to the movement of the diffusing gases. It will further be assumed that the diffusion characteristics of the waste cover are uniform with depth, so that a single diffusion coefficient (D) represents an average for the various soil layers.

$$\frac{\partial C}{\partial t} = D \frac{\partial^2 C}{\partial z^2} - \lambda_R C$$

$$\text{if } \frac{\partial C}{\partial t} = 0 \quad \text{then} \quad D \frac{\partial^2 C}{\partial z^2} = \lambda_R C$$

The source region is assumed to be a homogenous layer of soil with the same diffusion coefficient as the soil cover. In steady-state, the waste region has a constant gas concentration (C_0) in the pore space. Because this concentration is proportional to the release rate from the waste solids, it will change slowly with time, as discussed above.

The solution to the differential equation has an exponential dependence in elevation as shown below. The boundary conditions that the soil concentration is C_0 at the top of the waste ($z=0$) and zero at the ground surface ($z=z_0$) have been included. The solution is only valid from $z=0$ to $z=z_0$.

$$C(z) = C_0 \left[\frac{e^{-z\mu} - e^{-(2z_0 - z)\mu}}{1 - e^{-2z_0\mu}} \right] \quad \text{and} \quad \mu = \sqrt{\frac{\lambda_R}{D}}$$

$$J(z) = -D \frac{\partial C}{\partial z} \quad \text{thus} \quad J(z) = C_0 \mu D \frac{e^{-z\mu} + e^{-(2z_0 - z)\mu}}{1 - e^{-2z_0\mu}}$$

$$J(0) = C_0 \mu D \frac{1 + e^{-2z_0\mu}}{1 - e^{-2z_0\mu}} \quad \text{and} \quad J(z_0) = \frac{2C_0 \mu D e^{-z_0\mu}}{1 - e^{-2z_0\mu}}$$

where

- $C(z)$ = concentration of a gaseous radionuclide in the soil at elevation z in the soil above the waste, in Ci per m^3 of soil. At the bottom of the soil cover, the soil concentration matches the gas concentration in the waste, i.e., $C=C_0$. At the top of the cover the gas concentration is zero.
- C_0 = average gaseous radionuclide concentration in the waste pore space available for diffusion, in Ci per m^3 of waste. Note that this is the total gaseous activity in the waste pore space divided by the total volume of the waste.
- D = diffusion coefficient for low atomic number gases moving through waste and soil, $0.01 \text{ cm}^2/\text{s} = 31.56 \text{ m}^2/\text{y}$
- $J(z)$ = upward diffusion flux at elevation z above the waste, in Ci/m^2 per y
- z = vertical position in the soil, in m. The bottom of the soil column is $z=0$, while the ground surface is $z=z_0=5 \text{ m}$.
- λ_R = radioactive decay constant for the nuclide, per y.
- μ = inverse length parameter characteristic of gaseous radionuclide with a particular half life diffusing in a particular medium, in m^{-1}

Earlier performance assessment calculations did not use the boundary condition that the surface concentration is zero. In effect, an infinite medium was assumed. This underestimates the surface diffusion flux (J) by at least a factor of two. Longer half-life nuclides show a larger difference. The 2001 ILAW PA included this boundary condition.

Only ^{222}Rn is an inert gas that will faithfully follow the diffusion model. The ^3H and ^{14}C will undergo chemical reactions and be part of compounds that are likely to exist in the soil or the waste corrosion products. For ^3H these compounds include water and hydroxides. For ^{14}C these compounds include carbonates and carbides. The effect of these chemical reactions is to slow the migration to the surface and the resulting release rate from the ground surface. This effect has not been included in the diffusion calculations to maximize estimated consequences.

If the decay half life is very long, the above equations simplify to the equations shown below. In steady-state, the diffusion flux is constant with elevation. These equations represent the diffusion of ^3H and ^{14}C very well due to the long half lives.

$$C(z) = C_0 \left(1 - \frac{z}{z_0} \right) \quad \text{and} \quad J(z) = \frac{C_0 D}{z_0}$$

The calculation of gaseous radionuclide flux at the ground surface requires a value for the steady state concentration in the waste matrix (C_0). Since the concentration available for diffusion (C_0) varies slowly with time, the rate of addition of gas to the pore space equals the rate of removal. This can be stated mathematically as shown below. The first equation defines "steady state". The second equation says that the rate at which gaseous radionuclides are entering the waste pore space is equal to the rate at which they decay and diffuse. In this simple one-dimensional model the diffusion is either up or down. Hence, the total loss is double the diffusion flux in one direction. In the final equation for steady state concentration the numerator is the rate of addition of gaseous nuclides to the waste pore space.

$$\text{rate of addition} = \text{rate of removal}$$

$$\lambda_C Q_S = \lambda_R Q_0 + 2 A J(0) \quad \text{and} \quad J(0) = \frac{Q_0 \mu D}{A z_W} \Omega$$

$$\lambda_C Q_S = \lambda_R Q_0 + \frac{2 Q_0 \mu D}{z_W} \Omega \quad \text{where} \quad \Omega = \frac{1 + e^{-2 z_0 \mu}}{1 - e^{-2 z_0 \mu}}$$

$$\text{finally, } C_0 = \frac{\lambda_C Q_S}{\lambda_R V + 2 A \mu D \Omega}$$

where

- A = surface area footprint of the waste disposal site, in m^2
- C_0 = average gaseous radionuclide concentration in the waste pore space available for diffusion, in Ci per m^3 of waste. Note that this is the total gaseous activity in the waste pore space divided by the total volume of the waste.
- D = diffusion coefficient for low atomic number gases moving through waste and soil, $0.01 \text{ cm}^2/\text{s} = 31.56 \text{ m}^2/\text{y}$
- $J(z)$ = upward diffusion flux at elevation z above the waste, in Ci/m^2 per y
- Q_0 = quantity of a gaseous radionuclide located in the waste pore space, in Ci
- Q_S = quantity of a gaseous radionuclide embedded in the waste solids, in Ci
- z_0 = thickness of the soil above the waste, 5 m

- z_W = thickness of the waste, in m. The volume of the waste is the product of the footprint area and the thickness: $V = A z_W$.
 λ_C = fraction of the total released from the waste solids into the waste pore space per year.
 λ_R = radioactive decay constant for the nuclide, per y
 μ = inverse length parameter characteristic of gaseous radionuclide with a particular half life diffusing in a particular medium, in m^{-1}

It should be noted that the above equation for gaseous radionuclide concentration in the pore space reduces to the expected form when the radioactive decay constant becomes very small. One way to derive the expected form is to write the no-decay flux in terms of both the source release rate and the surface release rate. By conservation of matter in a simple one-dimensional model, this condition is shown in the equation below. The factor of two in the first equation stems from the assumption that only half the released activity diffuses upward. The other half diffuses downward. This formula can be used to calculate the initial concentration in the waste. The final equation for diffusion flux at the ground surface when radioactive decay may be important is shown below also.

$$J = \frac{D C_0}{z_0} = \frac{\lambda_C Q_S}{2 A} \quad \text{thus} \quad C_0 = \frac{\lambda_C Q_S z_0}{2 A D}$$

Determining the initial concentration of ^{222}Rn is complicated by two sources of the gas. One is the release from the waste solids. The other is the decay of ^{226}Ra that has already been released from the waste solids. The ^{226}Ra produces ^{222}Rn by radioactive decay. The total amount of ^{226}Ra slowly increases for two reasons. First, an increasing fraction of the waste matrix has decomposed. Second, ^{226}Ra is being produced by the radioactive decay of ^{238}U and ^{234}U that have been released. The peak ^{222}Rn flux occurs after all the waste has decomposed, and the ^{226}Ra has reached radioactive equilibrium with the ^{238}U . This equilibrium occurs after times greater than 1×10^6 years after closure. Note that the uranium decay chain is assumed to remain in the waste disposal facility indefinitely. This ignores the mobility of uranium in the soil, but is conservative. For ^{222}Rn , the steady state concentration in the waste pore space is calculated as shown in the equation below. In the final equation for steady state concentration the numerator is the rate of addition of ^{222}Rn to the waste pore space.

$$\begin{aligned}
 &\text{rate of addition} = \text{rate of removal} \\
 &\lambda_C Q_{S,\text{Rn-222}} + \lambda_{\text{Rn-222}} Q_{0,\text{Ra-226}} = \lambda_{\text{Rn-222}} Q_{0,\text{Rn-222}} + 2 A J_{\text{Rn-222}}(0) \\
 &\text{finally, } C_{0,\text{Rn-222}} = \frac{\lambda_C Q_{S,\text{Rn-222}} + \lambda_{\text{Rn-222}} Q_{0,\text{Ra-226}}}{\lambda_{\text{Rn-222}} V + 2 A \mu D \Omega}
 \end{aligned}$$

If the initial inventory per unit area of ^{238}U in the disposal facility is 1 Ci/m^2 (i.e., $1 \text{ Ci } ^{238}\text{U}$ per m^2 of facility footprint), then after a few million years, the waste solids have decomposed (i.e. $Q_S=0$) and the ^{234}U , ^{230}Th , ^{226}Ra , and ^{222}Rn are all in radioactive equilibrium with the ^{238}U . Thus the ^{226}Ra inventory in the trench is 1 Ci/m^2 . It is assumed that no migration of the uranium out of the trench has taken place during this time. If the waste thickness ($z_W=V/A$) is 1 m, the resulting concentration of ^{222}Rn is 0.420 Ci/m^3 . The calculation of the diffusion flux for ^{222}Rn is carried out using the formula given previously for $J(z_0)$. The ^{222}Rn flux at the ground surface

through 5 m of cover with a diffusion coefficient of $0.01 \text{ cm}^2/\text{s}$ is 0.0275 Ci/m^2 per year, or 671 pCi/m^2 per second. Note that the performance objective for radon emanation is 20 pCi/m^2 per second.

E.2. MECHANISMS AFFECTING THE DIFFUSION FLUX

There are four environmental factors that can affect the rate of gas transport through the soil. These are barometric pressure, temperature, wind, and rain (Jury 1991). Each of these will be discussed in terms of its effect on the gaseous diffusion rate from the ground surface calculated above.

E.2.1. Atmospheric Pressure

An increase in atmospheric pressure will compress the air above the soil and drive air into the soil. The motion of air into the soil is impeded by the tortuosity of the flow channels and the small diameter of the surface openings between soil grains. When the atmospheric pressure decreases gases in the soil expand and the gas near the surface is released into the atmosphere. The effect on gases diffusing from underground is to slow the release rate while the atmospheric pressure increases, and increase the release rate when the atmospheric pressure decreases.

However, atmospheric pressure variation represents a very small change in the average pressure. The annual standard deviation of barometric pressure variations was at most 0.72% during the years 1988 through 1991 (WHC-EP-0651). The annual extremes for these years were at most 3.1% from the average. Such small variation suggests the influence of atmospheric pressure changes is small. This suspicion is born out in literature cited in *Soil Physics* (Jury 1991). A soil permeable column 3 m deep was affected by typical barometric pressure variations to a depth of at most 0.56 cm. Thus, barometric pressure changes will have little influence on the average diffusion flux from the surface.

E.2.2. Atmospheric Temperature

The air temperature at the ground surface varies sinusoidally during the day as well as during the year. The daily variation does not penetrate as deeply as the seasonal variation. The seasonal temperature change at a depth of 5 m in sandy soil is about 10% of the seasonal temperature change at the surface (Jury 1991). The variation in temperature decreases exponentially with distance below the surface. In addition, there is a phase shift that increases with depth. The cyclic variation in soil temperature will cause air movement due to the expansion and contraction of the air as the temperature increases and decreases. This air movement can be expected to increase the average diffusion flux at the surface.

However, the influx of air is limited to a shallow layer near the surface, just as with barometric pressure changes. The monthly average air temperature at Hanford typically ranges from -12 C to 30 C with an average around 12 C (PNL-9809). Thus the near-surface soil layer experiences a seasonal temperature variation no more than 10% on the absolute temperature scale. While this variation is an order of magnitude greater than the barometric pressure change, it is confined to a relatively shallow layer near the surface. Thus, temperature changes will have little effect on the average diffusion flux from the surface. If the 10% variation is assumed to

accelerate gas transport in the top 10% of the soil cover, then the soil cover thickness is effectively reduced by 0.5 m. With a cover thickness of 4.5 m, the ^{222}Rn diffusion flux increases from 0.96 pCi/m^2 per second to 1.98 pCi/m^2 per second.

E.2.3. Wind

The motion of air over the ground surface will create areas with higher pressure and areas with lower pressure. This varying surface pressure is similar in effect to the barometric pressure variation, but on a much shorter time scale. Experience with temperature changes suggests that the rapid variation will not penetrate as deeply into the soil. Thus the enhanced gaseous release rate will affect gas migration to a depth less than the 0.5 cm predicted for barometric pressure variation. It is concluded that wind effects will have little effect on the average diffusion flux from the surface.

C.2.4. Rainfall

As the water infiltrates the soil, it displaces some air from the soil pore space. As the water moves deeper, the pore space is refilled with air from the nearby pore space. This air motion will increase the mixing between adjacent layers and should increase the average diffusion flux as well.

However, as the water moves deeper in the soil, the water and air coexist in the pore space so that the motion of water has no effect on the diffusion of gases. Thus the rainfall effect is limited to a shallow surface layer. The overall effect on the average diffusion flux is very small. The largest recorded 24-h rainfall at the Hanford Meteorological Station is 4.85 cm (1.91 in), which occurred on October 1 and 2, 1957. Using a soil porosity of 35% means the water would saturate the soil to a depth of 14.0 cm (5.5 in). Thus the normal rainfall at Hanford affects gas motion in the soil to a depth considerably less than the normal temperature variation.

E.3. DOSE CONSEQUENCES FROM AIR EMISSIONS

Two cases are described in the "No Infiltration Scenarios" of Table 2 (of the main text). The first is the "Offsite Farmer" who is downwind of the disposal site. The second is the "Onsite Resident" who lives in a dwelling affected by the gaseous emission. These scenarios are presented in more detail below. In addition, the two are compared to show that the offsite dose is always lower than the onsite dose.

The H-3 is normally found as water and would be released as water vapor. The C-14 is assumed bound in an organic compound. Presumably the organic compound could be volatile as well. The most likely carbon compound to be emitted is carbon dioxide. However, the inhalation dose factor for carbon dioxide is about 90 times smaller than for organically bound C-14, which has been used in this report. Finally, the inhalation dose from radon compounds depends on the ease with which the particulate progeny of the inert gas become attached to dust particles in the air, as well as the relative amounts of the short half life progeny. For this reason, the radon emission is limited to $20 \text{ pCi m}^{-2} \text{ s}^{-1}$ (RPP-13263 and RPP-14283).

The air pathway dose downwind was estimated using the formula shown below.

$$\text{Air Pathway Dose} = J A T DF$$

where

- J = flux of the radionuclide from the soil surface, Ci/m² per y
- A = cross-sectional area of the disposal facility trenches when viewed from above
- T = duration of the release, 1 y
- DF = air pathway dose factor from CAP88-PC for an annual release, mrem

The air transport factor and air pathway dose factors presented below originally appeared in Section 3.2.1 of *Exposure Scenarios and Unit Dose Factors for the Hanford Immobilized Low-Activity Tank Waste Performance Assessment* (HNF-SD-WM-TI-707 Revision 1). The air transport factor is a bounding value that applies to annual emission near the border of a large area source. The dose factors were obtained using the CAP88-PC software from EPA (EPA 402-B-92-001).

E.3.1. Offsite Farmer

Radioactivity released into the air is carried by the wind to locations some distance from the emission source. As the airborne material travels it is diluted, so that the potential doses are lessened. However, the radioactive material may settle on crops and pastures, leading to ingestion pathway doses as well as the initial inhalation dose as the plume passes the receptor location. The inhalation dose to an individual downwind is calculated using the formula shown below.

$$H_{bt} = Q_{emit} X/Q' BR DF_{inh}$$

where,

- H_{bt} = inhalation dose at a downwind location, mrem
- Q_{emit} = activity released into the air as a gas or respirable particles, Ci
- X/Q' = air transport factor, 1.0E-4 s/m³
- BR = breathing rate, 2.64E-4 m³/s
- DF_{inh} = inhalation dose factor for a nuclide emanating from the ground surface, in mrem per pCi inhaled. Values are given in Table A22.

Because the emission from the ground surface is assumed to take place over the course of a year, the parameters in the above are chosen to represent the annual average. Thus the annual average breathing rate is 0.95 m³/h, or 2.64E-4 m³/s. In addition, the activity released into the air is the total released over the course of one year. Finally, the air transport factor (1.0E-4 s/m³) is a bounding value for an annual emission (WHC-SD-WM-TI-616). It includes the normal variation in wind direction, wind speed, and atmospheric stability.

Inserting values for inhalation dose factor, and assuming a 1 Ci emission over the course of a year, leads to unit dose factors of 0.0025 mrem per Ci H-3, and 0.055 mrem per Ci C-14.

An additional method for estimating the offsite dose that takes into account the ingestion pathways is the CAP88-PC software from EPA (EPA 402-B-92-001). **The software predicts that the total doses are 0.0237 mrem per Ci H-3, and 1.32 mrem per Ci C-14.** The program output is listed in the attachment to this appendix. Note that the air transport factor used by CAP88-PC is $7.98\text{E-}6 \text{ s/m}^3$ because a distance of 1000 m was input. The doses reported by CAP88-PC were then adjusted upward by the ratio of air transport factors, namely, $(1.0\text{E-}4)/(7.98\text{E-}6)=12.5$ to give the values reported above. An additional difference with the simple inhalation dose method is the assumed chemical form of the carbon. In CAP88-PC the carbon is in the form of carbon dioxide, while in the simple inhalation calculation, the organic form is used. Although the organic form of carbon leads to larger inhalation doses, the ingestion pathways in CAP88-PC give most of the dose. It should be noted that the various consumption parameters in CAP88-PC are larger. For example the fraction of contaminated vegetables is 100% in CAP88-PC. The value of CAP88-PC is that it can be used to demonstrate compliance with the 10 mrem/y performance objective for airborne emissions.

The CAP88-PC output (see attachment) was interpreted according to the following procedure. From the SYNOPSIS report, the worst-case wind direction was east southeast (ESE). From the CHI/Q TABLE report, the air transport factor for this direction is $7.980\text{E-}6 \text{ s/m}^3$. From the SUMMARY report, the effective dose equivalent for H-3 is 0.00189 mrem, and for C-14 0.105 mrem.

At a location where the air transport factor is $1.0\text{E-}4 \text{ s/m}^3$, the CAP88-PC dose is computed as shown below. These values are listed in Table C1.

$$\text{H-3:} \quad (0.00189 \text{ mrem})(1.0\text{E-}4 \text{ s/m}^3)/(7.98\text{E-}6 \text{ s/m}^3) = 0.0237 \text{ mrem}$$

$$\text{C-14:} \quad (0.105 \text{ mrem})(1.0\text{E-}4 \text{ s/m}^3)/(7.98\text{E-}6 \text{ s/m}^3) = 1.32 \text{ mrem}$$

The EPA default parameter file was used as input to the CAP88-PC calculations rather than the Hanford-specific parameter file.

E.3.2. Onsite Resident

For modeling the effect of contaminants emanating from the soil into a ventilated building, the equilibrium air concentration is the product of the emanation rate (activity per unit area per unit time) and the floor area divided by the building ventilation rate (volume per unit time). The ratio of ventilation rate to floor area is characteristic of classes of building construction. Typical ratios are less than 0.003 m/s. The individual exposed to gaseous emanations from the disposal site is located in a ventilated dwelling. For a constant emanation rate with a constant ventilation rate, the concentration starts at zero and rises according to the formula shown below.

$$C = (C_{eq}) \cdot \{1 - \text{Exp}[-(F)(t)/(V)]\}$$

$$C_{eq} = (E)/(F/A)$$

Where,

$$C = \text{time-dependent air concentration inside the building, Ci/m}^3$$

C_{eq} = steady-state (or equilibrium) air concentration, Ci/m³

E = contaminant emanation rate, Ci m⁻² s⁻¹

A = building floor area, m²

F = building ventilation rate, m³/s

V = building volume, m³

The ratio F/A is characteristic of certain dwellings. Two examples are listed below. Typical values will need to be established for dose calculation purposes. For radon, the DOE uses an emanation limit. For gases such as tritium (as water vapor) and carbon-14, a dose calculation can be performed.

house: $F/A = (10 \text{ cfm})/(100 \text{ ft}^2) = 1/600 \text{ ft/s} = 5.08\text{E-}4 \text{ m/s}$

office: $F/A = (100 \text{ cfm})/(200 \text{ ft}^2) = 1/120 \text{ ft/s} = 2.54\text{E-}3 \text{ m/s}$

Using the smaller value to maximize the equilibrium air concentration, and assuming a unit emanation rate (1 pCi m⁻² s⁻¹) leads to an equilibrium air concentration of 2000 pCi/m³, or 2 pCi/L in the dirt-floor house.

The inhalation dose received by exposure to radionuclides emanating from the soil into a dwelling is the product of the steady-state air concentration, the volume of air inhaled during the year, and the inhalation dose factor as shown in the equation below. Any decay factor is expected to be nearly one, since the nuclides that contribute to this pathway (tritium, C-14, Rn-220 and Rn-222) are expected to see very little change in the emanation rate due to radioactive decay.

$$H_{be} = Q_{ab} C_{eq} D_{inh} T_{inh}$$

Where

H_{be} = inhalation dose from the emanation of gaseous nuclides during one year, in mrem.

Q_{ab} = quantity of air inhaled by the person while in the dwelling, in m³/y. The value used is 8,040 m³/y based on the sleeping and indoor data for the rural pasture scenario in Table A9.

C_{eq} = steady-state concentration of gaseous nuclides in the dwelling, in pCi/L.

D_{inh} = inhalation dose factor for a nuclide, in mrem per pCi inhaled. Values are given in Table A22.

T_{inh} = inhalation exposure time of the individual, 1 y.

For H-3 and C-14, the inhalation doses inside a house are calculated as shown below. They are based on a unit emanation rate, 1 pCi m⁻² s⁻¹.

H-3: $(8,040 \text{ m}^3/\text{y})(2000 \text{ pCi/m}^3)(9.60\text{E-}8 \text{ mrem/pCi})(1 \text{ y}) = 1.54 \text{ mrem}$

C-14: $(8,040 \text{ m}^3/\text{y})(2000 \text{ pCi/m}^3)(2.09\text{E-}6 \text{ mrem/pCi})(1 \text{ y}) = 33.6 \text{ mrem}$

E.3.3. Comparison of the Offsite and Onsite Individuals

A basic difference between the two dose calculations is the input radioactivity. For the offsite dose calculation, the total released into the air during one year must be estimated. For the onsite dose calculation, the rate at which activity enters the air per unit area of floor space is needed. To put the two on a comparable level, assume the ground surface directly above the waste site has a surface area of 31,690 m². A uniform emanation rate of 1 pCi m⁻² s⁻¹ for one year will release 1 Ci of activity into the air.

$$(1 \text{ pCi m}^{-2} \text{ s}^{-1})(31,690 \text{ m}^2)(3.156 \times 10^7 \text{ s/y})(1 \text{ y}) = 1 \text{ Ci}$$

Because the offsite dose is smaller than the onsite dose for unit emission rates (Table E1), the waste site area needs to be increased to raise the offsite dose, which depends on the total emitted during the year. Table E1 shows the calculated doses and required disposal site area for the two doses to be equal. Because the disposal site surface area will be significantly less than a square kilometer (1.0E+6 m²), the onsite dose will always be greater than the offsite dose.

Table E1. Comparison of Onsite and Offsite Doses.

Volatile Nuclide	Onsite Resident (per pCi m ⁻² s ⁻¹)	Offsite Farmer (per Ci released)	Disposal Facility Surface Area
H-3	1.54 mrem	0.0237 mrem	2.06E+6 m ²
C-14	33.6 mrem	1.32 mrem	8.07E+5 m ²

Disposal Site Area is calculated as the Onsite dose multiplied by 31,690 m² and divided by the Offsite dose.

E.4. REFERENCES

- ENDF/B-VI, Evaluated Nuclear Data File, Release VI. This nuclear data library is maintained by the Cross Section Evaluation Working Group. Data and documentation are available from the National Nuclear Data Center, Brookhaven National Laboratory, Upton, New York. (also provided on www.nndc.bnl.gov).
- EPA 402-B-92-001, 1992, Parks, B. S., *User's Guide for CAP88-PC Version 1.0*, U.S. Environmental Protection Agency, Washington, D.C.
- HNF-SD-WM-TI-707, Revision 1, 1999, Rittmann, P. D., *Exposure Scenarios and Unit Dose Factors for the Hanford Immobilized Low-Activity Tank Waste Performance Assessment*, Fluor Federal Services, Inc., Richland, Washington.
- Jury, W. A., W. R. Gardner, W. H. Gardner, 1991, *Soil Physics*, John Wiley & Sons, Inc. New York, New York, chapter 6.
- PNL-9809, 1994, Hoitink, D. J, and K. W. Burk, *Climatological Data Summary 1993 with Historical Data*, Pacific Northwest National Laboratory, Richland, Washington.
- RPP-13263, Revision 0, Mann, F. M., 2002, *Performance Objectives for the Hanford Immobilized Low-Activity Waste (ILAW) Performance Assessment*, CH2M Hill Hanford Group, Inc., Richland, WA.
- RPP-14283, Revision 0, Mann, F. M., A. J. Knepp, J. W. Badden, and R. J. Puigh, 2003, *Performance Objectives for the Tank Farm Closure Risk Assessments*, CH2M Hill Hanford Group, Inc., Richland, WA.
- WHC-EP-0651, Revision 0, 1993, Crippen, M. D., *Barometric Pressure Variations*, Westinghouse Hanford Company, Richland, Washington.
- WHC-SD-WM-TI-616, Rittmann, P. D., 1994, *Dose Estimates for the Solid Waste Performance Assessment*, Westinghouse Hanford Company, Richland, WA.

C A P 8 8 - P C

Version 1.00

Clean Air Act Assessment Package - 1988

S Y N O P S I S R E P O R T

Non-Radon Individual Assessment
Nov 30, 1999 7:45 am

Facility: Various Cases
Address:
 City: Richland
 State: WA Zip: 99352

Effective Dose Equivalent
(mrem/year)

1.07E-01

At This Location: 1000 Meters East Southeast
Source Category:
 Source Type: Stack
 Emission Year: 1996

Comments:

Dataset Name: ILAW PA
Dataset Date: Nov 30, 1999 7:45 am
Wind File: WNDFILES\HB-200E.WND

Nov 30, 1999 7:45 am

SYNOPSIS
Page 1

MAXIMALLY EXPOSED INDIVIDUAL

Location Of The Individual: 1000 Meters East Southeast
Lifetime Fatal Cancer Risk: 2.61E-06

ORGAN DOSE EQUIVALENT SUMMARY

Organ	Dose Equivalent (mrem/y)
GONADS	4.68E-02
BREAST	1.19E-01
R MAR	2.08E-01
LUNGS	5.64E-02
THYROID	5.61E-02
ENDOST	4.32E-01
RMNDR	1.02E-01
EFFEC	1.07E-01

Nov 30, 1999 7:45 am

SYNOPSIS
Page 2

RADIONUCLIDE EMISSIONS DURING THE YEAR 1996

Nuclide	Class	Size	Source #1 Ci/y	TOTAL Ci/y
H-3	*	0.00	1.0E+00	1.0E+00
C-14	*	0.00	1.0E+00	1.0E+00

SITE INFORMATION

Temperature: 12 degrees C
Precipitation: 17 cm/y
Mixing Height: 1000 m

Nov 30, 1999 7:45 am

SYNOPSIS
Page 3

SOURCE INFORMATION

Source Number: 1

Stack Height (m): 1.00
Diameter (m): 0.10

Plume Rise
Pasquill Cat:

	A	B	C	D	E	F	G
Zero:	0.00	0.00	0.00	0.00	0.00	0.00	0.00

AGRICULTURAL DATA

	Vegetable	Milk	Meat
Fraction Home Produced:	1.000	1.000	1.000
Fraction From Assessment Area:	0.000	0.000	0.000
Fraction Imported:	0.000	0.000	0.000

Food Arrays were not generated for this run.
Default Values used.

DISTANCES USED FOR MAXIMUM INDIVIDUAL ASSESSMENT

1000

C A P 8 8 - P C

Version 1.00

Clean Air Act Assessment Package - 1988

D O S E A N D R I S K E Q U I V A L E N T S U M M A R I E S

Non-Radon Individual Assessment
Nov 30, 1999 7:45 am

Facility: Various Cases
Address:
 City: Richland
 State: WA Zip: 99352

Source Category:
 Source Type: Stack
 Emission Year: 1996

Comments:

Dataset Name: ILAW PA
Dataset Date: Nov 30, 1999 7:45 am
 Wind File: WNDFILES\HB-200E.WND

Nov 30, 1999 7:45 am

SUMMARY
Page 1

ORGAN DOSE EQUIVALENT SUMMARY

Organ	Selected Individual (mrem/y)
GONADS	4.68E-02
BREAST	1.19E-01
R MAR	2.08E-01
LUNGS	5.64E-02
THYROID	5.61E-02
ENDOST	4.32E-01
RMNDR	1.02E-01
EFFEC	1.07E-01

PATHWAY EFFECTIVE DOSE EQUIVALENT SUMMARY

Pathway	Selected Individual (mrem/y)
INGESTION	1.06E-01
INHALATION	2.82E-04
AIR IMMERSION	0.00E+00
GROUND SURFACE	0.00E+00
INTERNAL	1.07E-01
EXTERNAL	0.00E+00
TOTAL	1.07E-01

Nov 30, 1999 7:45 am

SUMMARY
Page 2

NUCLIDE EFFECTIVE DOSE EQUIVALENT SUMMARY

Nuclide	Selected Individual (mrem/y)
H-3	1.89E-03
C-14	1.05E-01
TOTAL	1.07E-01

Nov 30, 1999 7:45 am

SUMMARY
Page 3

CANCER RISK SUMMARY

Cancer	Selected Individual Total Lifetime Fatal Cancer Risk
LEUKEMIA	6.60E-07
BONE	7.65E-08
THYROID	2.55E-08
BREAST	4.67E-07
LUNG	2.80E-07
STOMACH	2.48E-07
BOWEL	1.26E-07
LIVER	2.70E-07
PANCREAS	1.67E-07
URINARY	8.36E-08
OTHER	2.04E-07
TOTAL	2.61E-06

PATHWAY RISK SUMMARY

Pathway	Selected Individual Total Lifetime Fatal Cancer Risk
INGESTION	2.60E-06
INHALATION	7.62E-09
AIR IMMERSION	0.00E+00
GROUND SURFACE	0.00E+00
INTERNAL	2.61E-06
EXTERNAL	0.00E+00
TOTAL	2.61E-06

Nov 30, 1999 7:45 am

SUMMARY
Page 4

NUCLIDE RISK SUMMARY

Nuclide	Selected Individual
	Total Lifetime Fatal Cancer Risk
H-3	5.12E-08
C-14	2.56E-06
TOTAL	2.61E-06

Nov 30, 1999 7:45 am

SUMMARY
Page 5INDIVIDUAL EFFECTIVE DOSE EQUIVALENT RATE (mrem/y)
(All Radionuclides and Pathways)

Distance (m)	
<hr/>	
Direction	1000
<hr/>	
N	3.2E-02
NNW	4.3E-02
NW	4.2E-02
WNW	3.4E-02
W	2.7E-02
WSW	2.0E-02
SW	2.1E-02
SSW	1.9E-02
S	2.4E-02
SSE	3.1E-02
SE	6.2E-02
ESE	1.1E-01
E	7.5E-02
ENE	4.3E-02
NE	3.1E-02
NNE	2.6E-02

Nov 30, 1999 7:45 am

SUMMARY
Page 6INDIVIDUAL LIFETIME RISK (deaths)
(All Radionuclides and Pathways)

Distance (m)	
Direction	1000
N	7.9E-07
NNW	1.0E-06
NW	1.0E-06
WNW	8.2E-07
W	6.6E-07
WSW	5.0E-07
SW	5.2E-07
SSW	4.8E-07
S	5.8E-07
SSE	7.6E-07
SE	1.5E-06
ESE	2.6E-06
E	1.8E-06
ENE	1.1E-06
NE	7.6E-07
NNE	6.4E-07

C A P 8 8 - P C

Version 1.00

Clean Air Act Assessment Package - 1988

C H I / Q T A B L E S

Non-Radon Individual Assessment
Nov 30, 1999 7:45 am

Facility: Various Cases
Address:
City: Richland
State: WA Zip: 99352

Source Category:
Source Type: Stack
Emission Year: 1996

Comments:

Dataset Name: ILAW PA
Dataset Date: Nov 30, 1999 7:45 am
Wind File: WNDFILES\HB-200E.WND

Nov 30, 1999 7:45 am

CHIQ
Page 1GROUND-LEVEL CHI/Q VALUES FOR H-3
CHI/Q TOWARD INDICATED DIRECTION (SEC/CUBIC METER)

Distance (meters)	
<hr/>	
Dir	1000
<hr/>	
N	2.429E-06
NNW	3.192E-06
NW	3.166E-06
WNW	2.521E-06
W	2.025E-06
WSW	1.524E-06
SW	1.595E-06
SSW	1.455E-06
S	1.784E-06
SSE	2.326E-06
SE	4.672E-06
ESE	7.980E-06
E	5.598E-06
ENE	3.251E-06
NE	2.330E-06
NNE	1.946E-06

Nov 30, 1999 7:45 am

CHIQ
Page 2GROUND-LEVEL CHI/Q VALUES FOR C-14
CHI/Q TOWARD INDICATED DIRECTION (SEC/CUBIC METER)

Distance (meters)	
<hr/>	
Dir	1000
<hr/>	
N	2.429E-06
NNW	3.192E-06
NW	3.166E-06
WNW	2.521E-06
W	2.025E-06
WSW	1.524E-06
SW	1.595E-06
SSW	1.455E-06
S	1.784E-06
SSE	2.326E-06
SE	4.672E-06
ESE	7.980E-06
E	5.598E-06
ENE	3.251E-06
NE	2.330E-06
NNE	1.946E-06

This page is intentionally left blank.

APPENDIX F
QUALITY ASSURANCE CONSIDERATIONS

This page is intentionally left blank.

Desk Instruction 2.0, Rev. 1

CHECKLIST FOR TECHNICAL PEER REVIEW

Document Reviewed - _____

Title: HNF-SD-WM-TI-707, Revision 2, *Exposure Scenarios and Unit Dose Factors for Hanford Tank Waste Performance Assessments*, April 2003

Author: Paul D. Rittmann

<u>Yes</u>	<u>No*</u>	<u>NA</u>	
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Referenced analyses appropriate.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Problem completely defined and all potential configurations considered.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Accident scenarios developed in a clear and logical manner.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Necessary assumptions explicitly stated and supported.
<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	Computer codes and data files documented.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Data used in calculations explicitly stated in document.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Data checked for consistency with original source information as applicable. <i>(Spot check)</i>
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Mathematical derivations checked including dimensional consistency of results. <i>(checked all non-decay equations, spot checked decay equations)</i>
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Models appropriate and used within range of validity, or use outside range of established validity justified.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Hand calculations checked for errors. Spreadsheet results should be treated exactly the same as hand calculations. <i>(Spot check)</i>
<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	Software input correct and consistent with document reviewed.
<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	Software output consistent with input and with results reported in document reviewed.
<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	Limits/criteria/guidelines applied to analysis results are appropriate and referenced. Limits/criteria/guidelines checked against references.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Safety margins consistent with good engineering practices.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Conclusions consistent with analytical results and applicable limits.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Results and conclusions address all points required in the problem statement.
<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	Format consistent with applicable guides or other standards.
<input type="checkbox"/>	<input type="checkbox"/>	<input checked="" type="checkbox"/>	** Review calculations, comments, and/or notes are attached.
<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	Document approved (for example, the reviewer affirms the technical accuracy of the document).

Robert Marusich / Robert Marusich
 Technical Peer Reviewer (printed name and signature)

4/2/03
 Date

* All "no" responses must be explained below or on an additional sheet.

** Any calculations, comments, or notes generated as part of this review should be signed, dated and attached to this checklist. The material should be labeled and recorded in such a manner as to be understandable to a technically qualified third party.

This page is intentionally left blank.